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## Keramamides E, G, H, and J, New Cyclic Peptides Containing an Oxazole or a Thiazole Ring from a *Theonella* Sponge

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Abstract. Four new cyclic peptides, keramamides E(1), G(2), H(3), and J(4), containing an oxazole or a thiazole ring have been isolated from the Okinawan marine sponge *Theonella* sp. and the structures elucidated by 2D NMR data and degradation experiments. The sequence of amino acid residues in  $1 \sim 4$  was determined on the basis of FAB MS/MS data.

Marine sponges of the genus *Theonella* have been shown to be a rich source of unique secondary metabolites with intriguing structures and interesting biological activities.<sup>1</sup> In our continuing studies on bioactive substances from marine organisms,<sup>2</sup> we previously isolated some cyclic peptides, keramamides  $A^3$ ,  $B \sim D^4$ , and  $F^5$ , from an Okinawan marine sponge *Theonella* sp. Further investigation on extracts of this sponge led to isolation of four new cyclic peptides, keramamides E (1), G (2), H (3), and J (4), containing an oxazole or a thiazole ring and many unusual amino acids. This paper describes the isolation and structure elucidation of  $1 \sim 4$  on the basis of spectral data, especially FAB MS/MS data, and chemical means.

The methanol/toluene (3:1) extract of the sponge collected off Kerama Islands, Okinawa, was partitioned between toluene and water. The chloroform extract of the aqueous phase was subjected to silica gel and Sephadex LH-20 columns, and HPLC on ODS to yield keramamides E (1, 4.6 x 10<sup>-5</sup> %, wet weight of the sponge), G (2, 5.3 x 10<sup>-5</sup> %), H (3, 2.8 x 10<sup>-5</sup> %), and J (4, 5.0 x 10<sup>-5</sup> %) together with known cyclic peptides, keramamides B (5)<sup>4</sup>, C (6)<sup>4</sup>, D (7)<sup>4</sup>, and F (8)<sup>5</sup>.

The molecular formula of keramamide E (1) was determined as  $C_{53}H_{75}N_{10}O_{12}Br$  by HRFABMS [m/z 1123.4818, (M+H)<sup>+</sup>,  $\Delta$  +1.0 mmu].<sup>6</sup> Amino acid analyses of the acid hydrolysate of 1 showed the presence of 1 mol each of proline (Pro), ornithine (Orn), isoleucine (Ile), alanine (Ala), and norvaline (nVal). Extensive analyses of the <sup>1</sup>H (Table 1) and <sup>13</sup>C NMR data of 1 including <sup>1</sup>H-<sup>1</sup>H COSY, HMQC, and NOESY spectra by comparison with those of keramamide B (5) suggested the gross structure of 1 containing a 2-hydroxy-3-methylpentanoic acid (Hmp), a 2-bromo-5-hydroxytryptophan (BhTrp), and segments a and b. The presence of an Hmp group at Ile-NH was revealed by the NOESY correlation between Ile-NH and Hmp- $\alpha$ . The <sup>1</sup>H and <sup>13</sup>C NMR spectral data of 1 [ $\delta$ H 8.22 (1H, s);  $\delta$ C 136.9 s, 139.5 d, and 165.2 s] indicated the presence of an oxazole ring in segment b. The sequence of amino acid residues in 1 was determined by FAB MS/MS experiments. The MS/MS product ions observed in the FAB MS/MS spectra from the protonated molecular ions (m/z 1123 and 1125) are shown in Table 2 and Fig. 1. Chiral HPLC analysis (SUMICHIRAL OA-5000) of the acid hydrolysate of 1 clarified that Ala, Ile, Orn, Pro, and nVal were L-form. The BhTrp residue was converted into Asp by treatment of 1 with ozone followed by

CH<sub>3</sub>CO<sub>3</sub>H,<sup>4</sup> while the  $\alpha$ -keto- $\beta$ -amino acid (a) was transformed into Leu by treatment of 1 with H<sub>2</sub>O<sub>2</sub>/NaOH.<sup>5</sup> Both Asp and Leu in the degradation products were determined to be L by the chiral HPLC analysis. The absolute stereochemistry of Hmp generated by alkaline hydrolysis of 1 was determined to be 2S,3S by comparison of retention times in chiral HPLC with those of (2S,3S)-, (2R,3R)-, (2S,3R)-, and (2R,3S)-Hmp derived from L-Ile, D-Ile, L-allo-Ile, and D-allo-Ile through deamination with NaNO<sub>2</sub>, respectively.<sup>7</sup> Thus the structure of keramamide E was concluded to be 1.

Similarly the absolute configurations of Hmp in keramamides  $B \sim D (5 \sim 7)^4$  which had remained to be determined were concluded to be 25.35 by the same method as described above.

The HRFABMS  $[m/2\ 921.3886\ (M+H)^+, \Delta\ -3.7\ mmu]$  of keramamide G (2) established the molecular formula to be  $C_{43}H_{56}N_{10}O_{11}S$ , which was the same as that of keramamide F (8). Amino acid analyses of the acid hydrolysate of 2 revealed 1 mol each of Ala, isoserine (Ise), 2,3-diaminopropionic acid (Dpr), and Ile. Extensive analyses of the  $^1H$  (Table 1) and  $^{13}C$  NMR data of 2 including  $^1H^{-1}H$  COSY, HMQC, and

Table 1. <sup>1</sup>II NMR Data of Keramamides E (1), G (2), II (3), and J (4) in DMSO-d<sub>6</sub>

osit	on	1 J (Hz)			2 J (Hz)	J(Hz)	4 J (Hz)
Imp	ОН	5.50 <sup>a</sup> (brd) 5.4	СНО		8.03(s)	8.03 (s)	8.03 (s)
ппр	α	3.74 (m)					
	В	1.70 (m)	Ise	NH	7.97 (m)	7.97 (m)	7.98 (m)
	y-CH <sub>2</sub>	1.34, 1.14 (m)		α	4.04 (m)	4.01 (m)	4.02 (m)
	y-CH <sub>1</sub>	0.82 (m)		β	3.14 (m)	3.14 (m)	3.13 (m)
	δ-CH <sub>3</sub>	0.81 (m)		•	3.50 (m)	3.48 (m)	3.43 (m)
	0 0113	0.01 (111)		OH	. ,	5.96 (d) 5.9	5.98 (d) 5.4
lie	NH	7.52 (d) 9.3					m <0 / 1
	α	4.26 (m)	Ile	NH	7.67 (d) 8.8	7.67 (d) 9.8	7.68 (d) 8.8
	β	1.70 (m)		α	4.24 (m)	4.25 (m)	4.25 (dd) 8.8, 6
	γ-CH <sub>2</sub>	1.40, 0.98 (m)		β	1.74 (m)	1.75 (m)	1.74 (m)
	ү-СН3	0.82 (m)		ү-СНЗ	0.85 (m)	0.83 (m)	0.85 (m)
	δ-CH <sub>3</sub>	0.81 (m)		ү-СН2	1.43 (m)	1.40 (m)	1.42 (m)
					1.05 (m)	1.00 (m)	1.05 (m)
nVal	NH	8.09 (d) 7.8		ъ-снз	0.85 (m)	0.85 (m)	0.84 (m)
	α	4.23 (m)					
	β	1.43, 1.53 (m)	Dpr	α-NH	8.08 (d) 7.8	8.12 (d) 7.8	8.14 (d) 6.8
	y	1.30 (m)		α	4.42 (m)	4.47 (m)	4.46 (m)
	δ	0.86 (m)		β	2.92 (m)	2.85 (m)	2.83 (m)
					3.60 (m)	3.57 (m)	3.46 (m)
Om	α-NH	7.90 (d) 6.8		B-NH	8.04 (m)	7.67 (m)	7.82 (m)
	α	4.45 (m)					
	β	1.63,1.36 (m)	Ala	NH	8.02 (m)	8.06 (d) 5.9	8.09 (d) 7.3
	γ	1.37 (m)		α	4.72 (m)	4.63 (m)	4.67 (m)
	δ	3.44, 2.64 (m)		β	1.24 (d) 6.8	1.36 (d) 6.8	1.34 (d) 7.3
	δ-NH	7.49 (m)					
			đ	2	6.91 (d) 15.1	6.69 (d) 15.1	6.61 (d) 14.9
Pro	α	4.34 (m)		3	7.42 (d) 15.1	7.15 (d) 15.1	7.15 (d) 14.9
	ß	2.17, 1.85 (m)		5	7.90 (s)	7.76 (s)	7.75 (s)
	γ	1.86, 1.47 (m)		9	4.81 (m)	4.76 (m)	4.75 (dd) 10.0, (
	δ	3.70, 3.51 (m)		10	9.31 (d) 6.8	9.25 (d) 7.3	9.28 (d) 6.8
			c	13	5.49 (m)	5.40 (dd) 9.5, 3.	
b	2	6.58 (d) 15.1		14 15	8.53 (d) 9.3 2.46 (m)	8.45 (d) 9.5 2.45 (m)	8.61 (d) 9.8 2.46 (m)
	3 5	7.14 (d) 15.1 8.22 (s)		16	0.75 (d) 6.8	0.79 (d) 7.3	0.77 (d) 7.3
	9	4.84 (m)		17	1.18 (m)	1.22 (m)	1.22 (m)
	10	9.20 (d) 5.4		18	0.84 (m)	0.89 (t) 7.3	0.82 (m)
a	13	4.89 (m)		19	3.96 (m)	3.93 (dd) 9.0, 5	
	14	8.18 (d) 5.0			4.02 (m)	4.02 (m)	3.99 (m)
	15	1.37 (m)		20	3.39 (s)	3.39 (s)	3.38 (s)
	16 17	1.46 (m) 0.83 (m)		α'		4.66 (m)	4.64 (m)
	18			β'	7.86 (s)	2.89 (m)	3.13 (m)
	19	0.83 (m) 1.49 (d) 6.8		р	7.00 (3)	3.35 (m)	3.35 (m)
	1 2	1.47 (0) 0.0		1'-NH	11.51 (s)	11.20 (8)	10.77 (s)
BhT	rp NH	8.12 (d) 11.2		2'	7.33 (d) 2.4		6.98 (d) 2.0
	α	4.58 (m)		4'	7.71 (d) 8.3	6.91 (d) 2.4	7.56 (d) 7.8
	ß	3.06, 2.83 (m)		5'	7.15 (t) 7.3		6.93 (t) 7.8
	1'	11.23 (s)		6'	7.10 (t) 7.3	6.59 (dd) 8.5, 2	
	4'	6.88 (d) 2.0		7'	7.41 (d) 7.8	7.04 (d) 8.5	7.28 (d) 7.8
	6'	6.58 (dd) 8.8, 2.0		α'-NH	9.07 (s)	8.08 (d) 9.0	8.05 (d) 7.0
	7'	7.02 (d) 8.8		5'-OH		8.76 (s)	
	OH	8.69 (s)					

a) 8 in ppm

NOESY spectra by comparison with those of keramamide F (8) showed the presence of a formyl group, an  $\alpha,\beta$ -dehydrotryptophan ( $\Delta$ -Trp), and segments c and d. The presence of a thiazole ring in segment d was

m/z	m/z	Assignment of MS/MS product ions <sup>c</sup>		
1123 <sup>a</sup>	1125 <sup>b</sup>	M + H (parent ion)		
1105	1107	M - H2O + H		
1067	1069	$M - C_4 H_9 + 2H$		
1037	1039	$M - C_5H_{1,1}O + 2H$		
1009	1011	lle-nVal-cyclo(Om-Pro-a-b-BhTrp) + 2H		
896	898	nVal-cyclo(Om-Pro-a-b-BhTrp) + 2H		
797	799	eyclo(Om-Pro-a-b-BhTrp) + 2H		
683	685	Pro-a-b-BhTrp + 2H		
655	657	Pro-a-b-BhTrp - CO + 2H		
403	403	Рто-а-h + Н		
327	327	Hmp-Ile-nVal + H		
306	306	a-b + H		
228	228	Hmp-lle		
200	200	Hmp-lle - CO		
165	165	b + H		
122	122	b - C <sub>2</sub> H <sub>5</sub> N		
115	115	Hmp		

Table 2. FAB MS/MS Data of Keramamide E (1)

a) $^{79}$ Br parent ion. b) $^{81}$ Br parent ion. c)The amide bond cleavages are assumed to occur between NH and CO (the B-type fragmentation).

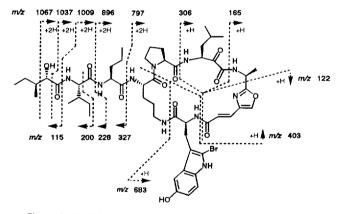


Fig. 1 FAB MS/MS Fragmentations of Keramamide E (1)

indicated by the  $^{1}$ H and  $^{13}$ C NMR spectral data of 2 [ $\delta_{H}$  7.90 (1H, s);  $\delta_{C}$  149.4 s, 123.1 d, and 165.9 s]. Further substantial evidences for the structure of 2 were obtained from the FAB MS/MS spectrum from the (M + H)<sup>+</sup> ion (m/z 921), which afforded MS/MS product ions corroborating well the amino acid sequence as shown in Fig. 2. Thus the gross structure of keramamide G (2) was elucidated to be the same as that of keramamide F (8). The differences between 2 and 8 were found only for carbon chemical shifts at C-13 (2,  $\delta_{C}$  56.7; 8,  $\delta_{C}$  59.7), C-16 (2,  $\delta_{C}$  14.3; 8,  $\delta_{C}$  16.0), and C-17 (2,  $\delta_{C}$  26.7; 8,  $\delta_{C}$  23.4), suggesting that the stereochemistry at C-13 of the  $\alpha$ -keto- $\beta$ -amino acid (c) in 2 was different from that of 8. Oxidation of 2 with  $H_{2}O_{2}/N_{3}O_{4}O_{5}$  followed by acid hydrolysis furnished IIe of which the absolute configuration was determined to be D by chiral HPLC analysis. Chiral HPLC analysis of the acid hydrolysate of 2 clarified that Ala, IIe, and Dpr were L-form, while Ise was D-form. Treatment of 1 with ozone for degradation of the (O-methylserine)thiazole (d) yielded O-methylserine, which was L-form from chiral HPLC analysis. Thus the structure of keramamide G was assigned to be 2.

Fig. 2 FAB MS/MS Fragmentations of Keramamide G (2)

The HRFABMS  $[m/z \ 1018.3190, M^+, \Delta -2.9 \text{ mmu}]$  of keramamide H (3) established its molecular formula to be  $C_{43}H_{57}N_{10}O_{12}BrS$ . Amino acid analyses of the acid hydrolysate of 3 revealed 1 mol each of Ala, Ise, Dpr, and Ile. The  $^1H$  NMR (Table 1) spectrum of 3 was almost the same as that of 2 except for its tryptophan moiety. Comparison of the  $^1H^{-1}H$  COSY spectrum of 3 with those of 1 and 2 revealed the presence of BhTrp  $[\delta_H \ 11.20 \ (s), 6.91 \ (d, J=2.4 \ Hz), 6.59 \ (dd, J=8.5 \ and 2.4 \ Hz), 7.04 \ (d, J=8.5 \ Hz), 8.08 \ (d, J=9.0 \ Hz), and 8.76 \ (s)] for 3 in place of <math>\Delta$ -Trp for 2. Furthermore, the FAB MS/MS spectra from the protonated molecular ions  $(m/z \ 1017 \ and \ 1019)$  of 3 supported the gross structure proposed for keramamide H. Ala, Ile, and Dpr in the acid hydrolysate of 3 were L-form by chiral HPLC analysis, while Ise was D-form. The BhTrp residue in 3 was converted into Asp by treatment of 3 with ozone followed by  $CH_3CO_3H$ , while the  $\alpha$ -keto- $\beta$ -amino acid (c) was transformed into Ile by treatment of 3 with  $H_2O_2/NaOH$ . Degradation of the (O-methylserine)thiazole (d) with ozone yielded O-methylserine. Asp, Ile, and O-methylserine in the degradation products were determined to be L by chiral HPLC analysis. Thus the structure of keramamide H was assigned to be 3.

The molecular formula of keramamide J (4) was determined as C<sub>43</sub>H<sub>58</sub>N<sub>10</sub>O<sub>11</sub>S by HRFABMS [m/z 923.4099 (M + H)<sup>+</sup>,  $\Delta$  +1.3 mmu] and the molecular weight was larger than that of keramamide G (2) by 2 Daltons. The <sup>1</sup>H NMR data (Table 1) of 4 were similar to those of 2 except for resonances of tryptophan moiety. Aromatic proton signals [ $\delta$ <sub>H</sub> 6.98 (d, J = 2.0 Hz), 7.56 (d, J = 7.8 Hz), 6.93 (t, J = 7.8 Hz), 7.02 (t, J = 7.8 Hz), 7.28 (d, J = 7.8 Hz), and 8.05 (d, J = 7.0 Hz)] for 4 were assigned as those of tryptophan by the <sup>1</sup>H-<sup>1</sup>H COSY data of 4 and comparison with the chemical shifts reported for a tryptophan residue in polydiscamide A.<sup>9</sup> Hydrolysis of 4 under mild acidic condition followed by amino acid analysis of the products revealed 1 mol each of Ala, Ise, Dpr, Ile, and Trp. The amino acid sequence of 4 was deduced from the FAB MS/MS data. Chiral HPLC analysis of the acid hydrolysate of 4 clarified that Ala, Ile, Dpr, and Trp were L-form, while Ise was D-form. The  $\alpha$ -keto- $\beta$ -amino acid (c) was transformed into Ile by treatment of 4 with H<sub>2</sub>O<sub>2</sub>/NaOH, while the ( $\theta$ -methylserine)thiazole (d) was converted into  $\theta$ -methylserine by the same method as described above. Both Ile and  $\theta$ -methylserine were determined to be L by chiral HPLC analysis. Thus the structure of keramamide J was concluded to be 4.

Keramamides E (1), G (2), H (3), and J (4) are new cyclic peptides containing an oxazole or a thiazole ring in addition to various unusual amino acids. Although there are many reports on cyclic peptides having oxazole and/or thiazole ring(s) from tunicates<sup>10</sup> and terrestrial microorganisms,<sup>11</sup> very few peptides containing conjugated oxazole or thiazole ring(s) have been isolated from natural origin. Such  $\alpha$ -keto- $\beta$ -amino acids contained in 1 ~ 4 have been reported for poststatin, a peptide from *Streptomyces viridochromogenes*<sup>12</sup>, while 2-hydroxy-3-methylpentanoic acid moiety has been found in peptides from a black yeast *Aureobasidium pullulans*. <sup>13</sup> Keramamide E (1) exhibited cytotoxicity against L1210 murine leukemia cells and KB human epidermoid carcinoma cells with IC<sub>50</sub> values of 1.60 and 1.55 µg/mL, respectively, while keramamides G (2), H (3), and J (4) showed weak cytotoxicity (IC<sub>50</sub>~10 µg/mL).

**Experimental Section** 

General Methods. Optical rotations were measured on a JASCO DIP-370 polarimeter. UV and IR spectra were obtained on JASCO Ubest-35 and JASCO IR report-100 spectrometers, respectively. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a JEOL EX-400 spectrometer in DMSO-d<sub>6</sub>. The 2.50 ppm resonance of residual CD<sub>3</sub>SOCHD<sub>2</sub> and 39.5 ppm of (CD<sub>3</sub>)<sub>2</sub>SO were used for <sup>1</sup>H and <sup>13</sup>C NMR spectra as internal references, respectively. Mass spectra were obtained on a JEOL JMS-HX/HX 110A tandem mass spectrometer by using glycerol as a matrix.

Collection, Extraction, and Isolation. The sponge *Theonella* sp. was collected off Kerama Island, Okinawa and was kept frozen until used. The toluene/MeOH (1:3, 2 L x 2) extract of the sponge (4 kg, wet weight) was suspended with 1M NaCl (1 L) and was extracted with toluene (600 mL x 2). The aqueous layer was extracted with CHCl<sub>3</sub> (800 mL x 2). The CHCl<sub>3</sub>-soluble material (2.1 g) was subjected to a silica gel column with gradient elution of MeOH (2-50%) in CHCl<sub>3</sub>. The fraction eluted with 15% MeOH in CHCl<sub>3</sub> was separated by a silica gel column with toluene/MeOH (98:2 to 70:30) to afford crude peptide fractions I and II. The fraction I was further purified by a Sephadex LH-20 column (MeOH) followed by reversed-phase HPLC [YMC-Pack AM-324 ODS, Yamamura Chemical, 10 x 250 mm; flow rate: 2.0 mL/min; eluent: MeOH/H<sub>2</sub>O/CF<sub>3</sub>CO<sub>2</sub>H, 70:30:0.1; UV detection at 254 nm] to give keramamides B (5, 7.4 mg, r<sub>R</sub> 48.0 min), C (6, 11.8 mg, r<sub>R</sub> 40.1 min), and D (7, 10.1 mg, r<sub>R</sub> 31.5 min) and a fraction (12.3 mg, r<sub>R</sub> 37.1 min), which was purified by the same reversed-phase HPLC column with CH<sub>3</sub>CN/H<sub>2</sub>O/CF<sub>3</sub>CO<sub>2</sub>H (42:58:0.1) to afford keramamide E (1, 1.8 mg, r<sub>R</sub> 63 min). The fraction II was separated by a Sephadex LH-20 column (MeOH) followed by the same HPLC column with CH<sub>3</sub>CN/H<sub>2</sub>O (40:60) to give keramamide F (8, 5.1 mg, r<sub>R</sub> 22.0 min) and the fractions III (r<sub>R</sub> 23 ~ 26 min) and IV (r<sub>R</sub> 26 ~ 37 min). Fraction III was subjected to the same HPLC column with MeOH/H<sub>2</sub>O/CF<sub>3</sub>CO<sub>2</sub>H (70:30:0.1) to afford keramamides G (2, 2.1 mg, r<sub>R</sub> 16 min) and H (3, 1.1 mg, r<sub>R</sub> 18 min). Fraction IV was purified under the same HPLC condition to yield keramamide J (4, 2.0 mg, r<sub>R</sub> 36 min).

Keramamide E (1): colorless solid;  $[\alpha]^{22}_D$  -39° (c 0.1, MeOH); IR (KBr)  $\nu_{max}$  3400, 1670, 1640, and 1520 cm<sup>-1</sup>; UV(MeOH)  $\lambda_{max}$  269 ( $\epsilon$  30900) and 310 (sh) nm; <sup>1</sup>H NMR (Table 1); <sup>13</sup>C NMR (DMSO- $d_6$ ) δ<sub>C</sub> 172.6 (Hmp-CO), 74.9 (Hmp- $\alpha$ ), 38.1 (Hmp- $\beta$ ), 23.2 (Hmp-γ-CH<sub>2</sub>), 15.3 (Hmp-γ-CH<sub>3</sub>), 11.7 (Hmp-δ-CH<sub>3</sub>), 170.6 (Ile-CO), 55.8 (Ile- $\alpha$ ), 37.3 (Ile- $\beta$ ), 24.2 (Ile-γ-CH<sub>2</sub>), 15.4 (Ile-γ-CH<sub>3</sub>), 10.9 (Ile-δ-CH<sub>3</sub>), 171.2 (nVal-CO), 52.0 (nVal- $\alpha$ ), 34.0 (nVal- $\beta$ ), 18.5 (nVal- $\gamma$ ), 13.6 (nVal- $\delta$ ), 169.4 (Orn-CO), 49.7 (Orn- $\alpha$ ), 37.8 (Orn- $\beta$ ), 25.1 (Orn- $\gamma$ ), 38.5 (Orn- $\delta$ ), 171.2 (Pro-CO), 58.4 (Pro- $\alpha$ ), 29.5 (Pro- $\beta$ ), 24.4 (Pro- $\gamma$ ), 46.9 (Pro- $\delta$ ), 163.9 (C-1), 123.5 (C-2), 127.4 (C-3), 136.9 (C-4), 139.5 (C-5), 165.2 (C-7), 43.9 (C-9), 159.6 (C-11), 195.8 (C-12), 52.4 (C-13), 29.1 (C-15), 24.8 (C-16), 20.8 (C-17), 23.1 (C-18), 18.0 (C-19), 170.7 (BhTrp-CO), 53.4 (BhTrp- $\alpha$ ), 27.9 (BhTrp- $\beta$ ), 109.5 (BhTrp-2'), 109.0 (BhTrp-3'), 102.3 (BhTrp-4'), 150.7 (BhTrp-5'), 111.5 (BhTrp-6'), 110.9 (BhTrp-7'), 128.2 (BhTrp-8'), and 130.5 (BhTrp-9'); FABMS (positive) m/z 1123 and 1125 (1:1) (M+H)+; exact mass found m/z 1123.4818, calcd for C<sub>53</sub>H<sub>76</sub>N<sub>10</sub>O<sub>12</sub>Br 1123.4808; FAB MS/MS (Table 2).

Keramamide G (2): colorless solid;  $[\alpha]^{21}_D + 10^\circ$  (c 0.12, MeOH); IR (KBr)  $v_{max}$  3400, 1660, and 1520 cm<sup>-1</sup>; UV(MeOH)  $\lambda_{max}$  224 (ε 26500), 279 (21200), and 340 (11700) nm; <sup>1</sup>H NMR (Table 1); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>) δ<sub>C</sub> 161.5 (CHO), 171.4 (Ise-CO), 70.2 (Ise-α), 41.6 (Ise-β), 170.1 (Ile-CO), 56.2 (Ile-α), 37.1 (Ile-β), 15.3 (Ile-γ-CH<sub>3</sub>), 24.3 (Ile-γ-CH<sub>2</sub>), 11.0 (Ile-δ-CH<sub>3</sub>), 169.3 (Dpr-CO), 51.1 (Dpr-α), 41.6 (Dpr-β), 173.8 (Ala-CO), 47.6 (Ala-α), 20.0 (Ala-β), 164.7 (Δ-Trp-CO), 121.9 (Δ-Trp-α), 124.0 (Δ-Trp-CO)

β), 127.1 (Δ-Trp-2'), 109.6 (Δ-Trp-3'), 118.0 (Δ-Trp-4'), 121.9 (Δ-Trp-5'), 120.0 (Δ-Trp-6'), 111.9 (Δ-Trp-7'), 135.4 (Δ-Trp-8'), 127.3 (Δ-Trp-9'), 163.3 (C-1), 124.0 (C-2), 132.1 (C-3), 149.4 (C-4), 123.1 (C-5), 165.9 (C-7), 53.7 (C-9), 161.0 (C-11), 197.2 (C-12), 56.7 (C-13), 37.0 (C-15), 14.3 (C-16), 26.7 (C-17), 11.6 (C-18), 72.6 (C-19), and 58.3 (C-20); FABMS (positive) m/z 921 (M+H)+; exact mass found m/z 921.3886, calcd for C<sub>43</sub>H<sub>57</sub>N<sub>10</sub>O<sub>11</sub>S 921.3923; FAB MS/MS (positive) m/z 921 [M + H]+, 903 [M - H<sub>2</sub>O + H]+, 877 [M - CH<sub>2</sub>NO + H]+, 833 [M - C<sub>3</sub>H<sub>6</sub>NO<sub>2</sub> + H]+, 806 [Ile-cyclo(Dpr-Ala-c-d- $\Delta$ Trp) + 2H]+, 790 [Ile-cyclo(Dpr-Ala-c-d- $\Delta$ Trp) - NH + H]+, 693 [cyclo(Dpr-Ala-c-d- $\Delta$ Trp) + 2H]+, 677 [cyclo(Dpr-Ala-c-d- $\Delta$ Trp) - NH + H]+, 395 [d- $\Delta$ Trp + H]+, 386 [Ise(CHO)-Ile-Dpr-Ala + H]+, 315 [Ise(CHO)-Ile-Dpr + H]+, 229 [Ise(CHO)-Ile]+, 185 [ $\Delta$ Trp + H]+, and 117 [Ise(CHO) + H]+.

Keramamide II (3): colorless solid;  $[\alpha]^{20}_D$  -42° (c 0.055, MeOH); IR (KBr)  $v_{max}$  3400, 1660, and 1520 cm<sup>-1</sup>; UV (MeOH)  $\lambda_{max}$  223 (sh) and 277 ( $\epsilon$  23100) nm; <sup>1</sup>H NMR (Table 1); FABMS (positive) m/z 1017 and 1019 (1:1, M+H)<sup>+</sup>; exact mass found m/z 1018.3190 (M<sup>+</sup>), calcd for C<sub>43</sub>H<sub>57</sub>N<sub>10</sub>O<sub>12</sub><sup>81</sup>BrS 1018.3219; FAB MS/MS (positive) m/z (for <sup>79</sup>Br parent ion) and  $m/z^*$  (for <sup>81</sup>Br parent ion) 1017 and 1019\* [M + H]<sup>+</sup>, 999 and 1001\* [M - H<sub>2</sub>O + H]<sup>+</sup>, 973 and 975\* [M - CH<sub>2</sub>NO + H]<sup>+</sup>, 929 and 931\* [M - C<sub>3</sub>H<sub>6</sub>NO<sub>2</sub> + H]<sup>+</sup>, 902 and 904\* [Ile-cyclo(Dpr-Ala-c-d-BhTrp) + 2H]<sup>+</sup>, 886 and 888\* [Ile-cyclo(Dpr-Ala-c-d-BhTrp) - NH + H]<sup>+</sup>, 789 and 791\* [cyclo(Dpr-Ala-c-d-BhTrp) + 2H]<sup>+</sup>, 774 and 776\* [cyclo(Dpr-Ala-c-d-BhTrp) - NH + H]<sup>+</sup>, 703 and 705\* [Ala-c-d-BhTrp + H]<sup>+</sup>, 632 and 634\* [c-d-BhTrp + H]<sup>+</sup>, 491 and 493\* [d-BhTrp + H]<sup>+</sup>, 386 and 386\* [Ise(CHO)-Ile-Dpr-Ala + H]<sup>+</sup>, 315 and 315\* [Ise(CHO)-Ile-Dpr + H]<sup>+</sup>, 282 and 284\* [BhTrp + 2H]<sup>+</sup>, 229 and 229\* [Ise(CHO)-Ile]<sup>+</sup>, and 117 and 117\* [Ise(CHO) + H]<sup>+</sup>.

Keramamide J (4): colorless solid;  $[\alpha]^{18}_D$  +8.4° (c 0.1, MeOH); IR (KBr)  $v_{max}$  3410, 1680, and 1540 cm<sup>-1</sup>; UV (MeOH)  $\lambda_{max}$  222 (ε 45200) and 278 (27200) nm; <sup>1</sup>H NMR (Table 1); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>) δ<sub>C</sub> 161.4 (CHO), 171.4 (Ise-CO), 70.0 (Ise-α), 41.3 (Ise-β), 169.2 (Ile-CO), 58.3 (Ile-α), 37.1 (Ile-β), 15.3 (Ile-γ-CH<sub>3</sub>), 24.2 (Ile-γ-CH<sub>2</sub>), 11.0 (Ile-δ-CH<sub>3</sub>), 169.2 (Dpr-CO), 51.3 (Dpr-α), 41.9 (Dpr-β), 174.1 (Ala-CO), 48.1 (Ala-α), 19.7 (Ala-β), 171.6 (Trp-CO), 58.3 (Trp-α), 28.3 (Trp-β), 127.1 (Trp-2'), 111.3 (Trp-3'), 122.5 (Trp-4'), 118.1 (Trp-5'), 120.6 (Trp-6'), 111.5 (Trp-7'), 137.5 (Trp-8'), 127.3 (Trp-9'), 164.4 (C-1), 124.4 (C-2), 131.2 (C-3), 149.2 (C-4), 122.6 (C-5), 165.5 (C-7), 53.7 (C-9), 161.4 (C-11), 197.0 (C-12), 56.1 (C-13), 37.0 (C-15), 14.2 (C-16), 26.7 (C-17), 11.5 (C-18), 72.5 (C-19), and 58.3 (C-20); FABMS (positive) m/z 923 (M+H)+; exact mass found m/z 923.4099 calcd for C4<sub>3</sub>H<sub>5</sub>9N<sub>10</sub>O<sub>11</sub>S 923.4086; FAB MS/MS (positive) m/z 923 [M + H]+, 905 [M - H<sub>2</sub>O + H]+, 879 [M - CH<sub>2</sub>NO + H]+, 835 [M - C<sub>3</sub>H<sub>6</sub>NO<sub>2</sub> + H]+, 808 [Ile-cyclo(Dpr-Ala-c-d-Trp) + 2H]+, 792 [Ile-cyclo(Dpr-Ala-c-d-Trp) - NH + H]+, 695 [cyclo(Dpr-Ala-c-d-Trp) + 2H]+, 679 [cyclo(Dpr-Ala-c-d-Trp) - NH + H]+, 538 [c-d-Trp + H]+, 397 [d-Trp + H]+, 386 [Ise(CHO)-Ile-Dpr-Ala + H]+, 315 [Ise(CHO)-Ile-Dpr + H]+, 229 [Ise(CHO)-Ile-1]+ 188 [Trp + 2H]+ and 117 [Ise(CHO) + H]+

H]+, 329 [Ise(CHO)-Ile]+, 188 [Trp + 2H]+, and 117 [Ise(CHO) + H]+.

Amino Acid Analysis by Chiral IPLC. Keramamides E, G, or H (1~3, each 0.1 mg) was hydrolyzed with 6N HCl (1.0 mL) at 110°C for 24 h and keramamide J (4, 0.1 mg) was hydrolyzed with 4N methanesulfonic acid (0.1 mL) at 115°C for 24h. Chiral HPLC analyses were carried out using a SUMICHIRAL OA-5000 [Sumitomo Chemical Industry, 4 x 150 mm, 40 °C, UV detection at 254 nm]. Retention times (min) of standard amino acids were as follows: L-Orn (13.1) and D-Orn (14.2) [eluent: H<sub>2</sub>O containing 1.0 mmol/L of CuSO<sub>4</sub>, flow rate: 0.2 mL/min]; L-Ala (5.7), D-Ala (8.1), L-(O-Me)Ser (9.7), D-(O-Me)Ser (14.0), L-Pro (12.7), D-Pro (27.2), L-Dpr (19.7), D-Dpr (22.9), L-nVal (23.8), and D-nVal (44.1) [eluent: H<sub>2</sub>O containing 1.0 mmol/L of CuSO<sub>4</sub>, flow rate: 1.0 mL/min]; L-Ile (13.3), allo-L-Ile (11.3), D-Ile (21.3), and allo-D-Ile (17.4) [eluent: MeOH/H<sub>2</sub>O (15:85) containing 2.0 mmol/L of CuSO<sub>4</sub>, flow rate: 1.0 mL/min]; L-Ise (25.7), D-Ise (41.4), L-Trp (34.2), and D-Trp (37.0) [eluent: MeOH/H<sub>2</sub>O (30:70) containing 2.0 mmol/L of CuSO<sub>4</sub>, flow rate: 1.0 mL/min]. The retention times (min) of hydrolysates of 1 ~ 4 were as follows; L-Orn (13.1), L-Ala (5.7), L-Pro (12.7), L-nVal (23.8), and L-Ile (13.3) in 1; L-Ala (5.7), L-Dpr (19.7), L-Ile (13.3), and D-Ise (41.4) in 2 and 3; L-Dpr (19.7), L-Ile (13.3), L-Trp (34.2), and D-Ise (41.4) in

Preparation of 2-Hydroxy-3-methylpentanoic Acid. A solution of NaNO<sub>2</sub> (50 mg) in water (0.5 mL) was added to a stirred and ice-salt-cooled solution of L-Ile (50 mg) in 1N H<sub>2</sub>SO<sub>4</sub> (1 mL). The mixture was stirred for 12 h and then extracted with THF (10 mL). The THF solution was washed with brine, dried (MgSO<sub>4</sub>) and concentrated under reduced pressure to give (2S,3S)-Hmp (30 mg, 60%):  $^{1}$ H NMR (CDCl<sub>3</sub>)  $\delta$  0.94 (3H, t, J = 7.5 Hz), 1.03 (3H, d, J = 6.9 Hz), 1.29 (2H, m), 1.90 (1H, m), and 4.18 (1H,

d, J = 3.3 Hz). According to essentially the same procedure as described above, DL-Ile, allo-L-Ile, and allo-D-Ile afforded (2S,3S)- and (2R,3R)-Hmp, (2S,3R)-Hmp, and (2R,3S)-Hmp, respectively. (2S,3R)-Hmp: <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.89 (3H, d, J = 7.1 Hz), 0.97 (3H, t, J = 7.5 Hz), 1.38 (2H, m), 1.90 (1H, m), and 4.29 (1H, d, J = 2.7 Hz).

Determination of the Stereochemistry of 2-Hydroxy-3-methylpentanoic Keramamides E (1) and B ~ D (5 ~ 7). Keramamide E (1, 0.2 mg) was hydrolyzed with 1N NaOH/MeOH (1:4, 0.5 mL) at 50 °C for 8 h. The reaction mixture acidified by 1N HCI (0.3 mL) was then extracted with THF (5 mL x 3) and the residue was subjected to chiral HPLC analysis using a SUMICHIRAL OA-5000 | 4 x 150 mm, 30 °C, flow rate: 1.0 mL/min; eluent: CH<sub>3</sub>CN/H<sub>2</sub>O (20:80) containing 2.0 mmol/L of CuSO4; detection: UV at 254 nm]. Retention times (min) of synthetic 2-hydroxy-3-methylpentanoic acid (Hmp) isomers were as follows; 2S, 3R-Hmp (26.0), 2S, 3S-Hmp (28.9), 2R, 3S-Hmp (38.4), and 2R, 3R-Hmp (43.4). The retention time of Hmp in the hydrolysate of 1 was found to be 28.9 min. Keramamides B, C, or D (5 ~ 7, each 0.2 mg) was hydrolyzed with 1N NaOH/MeOH (1:4, 0.5 mL) at 50 °C for 8 h. The reaction mixture acidified by 1N HCI (0.3 mL) was then extracted with THF (5 mL x 3) and the residue was subjected to chiral HPLC analysis as described above. The retention times of Hmp in the hydrolysates of 5, 6, and 7 were found to be all 28.9 min.

Determination of the Stereochemistry of the (O-Methylserine)thiazole in Keramamides G (2), II (3), and J (4). A stream of O<sub>3</sub> was bubled into a 1 mL MeOH solution of keramamides G, H, or J (2-4, each 0.2 mg) at room temperature for 8 min. The reaction mixture was subjected to hydrolysis and the chiral HPLC analysis using a SUMICHIRAL OA-5000 [4 x 150 mm, 40 °C, flow rate: 1.0 mL/min; eluent: H<sub>2</sub>O containing 1.0 mmol/L of CuSO<sub>4</sub>]. Retention times of authentic L-(O-Me)Ser and D-(O-Me)Ser were 9.7 and 14.0 min, respectively. The retention times of (O-Me)Ser in the oxidation product of 2, 3, and 4 were found to be all 9.7 min.

Determination of the Stereochemistry of the α-Keto-β-amino Acid in Keramamides E (1), G (2), H (3), and J (4). To a stirred solution of keramamide E (1, 0.2 mg) in 5% NaOH (0.5 mL) was added dropwise 30% H<sub>2</sub>O<sub>2</sub> (0.1 mL). After stirring at 65 °C for 40 min the reaction mixture was hydrolyzed with 6N HCl (1.0 mL) at 110 °C for 24h and subjected to chiral HPLC analysis using SUMICHIRAL OA-5000 [40°C; flow rate: 1.0 mL/min; detection: UV at 254 nm]. Retention times of authentic L-Leu, D-Leu, L-Ile, D-Ile, allo-L-Ile, and allo-D-Ile [eluent: MeOH/H<sub>2</sub>O (15:85) containing 2.0 mmol/L of CuSO<sub>4</sub>] were 14.4, 21.9, 13.3, 21.3, 11.3, and 17.4 min, respectively. The retention time of Leu in the degradation product of 1 was found to be 14.4 min. Keramamides G, H, or J (2 ~ 4, each 0.2 mg) was treated with 30% H<sub>2</sub>O<sub>2</sub> (0.1 mL) in 5% NaOH (0.5 mL) at 65 °C for 40 min. The reaction mixture was hydrolyzed with 6N HCl (1.0 mL) at 110 °C for 24 h. The hydrolysate was applied to the chiral HPLC analysis as described above. The retention times of Ile in the degradation products of 2, 3, and 4 were found to be 21.3, 13.3, and 13.3 min, respectively.

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## References

- Fusetani, N.; Matsunaga, S.; Matsumoto, H.; Takebayashi, Y. J. Am. Chem. Soc. 1990, 112, 7053-7054 and references cited therein.
- Kobayashi, J.; Kondo, K.; Ishibashi, M.; Wälchli, M. R.; Nakamura, T. J. Am. Chem. Soc. 1993, 115, 6661-6665 and references cited therein.
- Kobayashi, J.; Sato, M.; Ishibashi, M.; Shigemori, H.; Nakamura, T.; Ohizumi, Y. J. Chem. Soc., Perkin Trans. 1. 1991, 2609-2611.
- Kobayashi, J.; Itagaki, F.; Shigemori, H.; Ishibashi, M.; Takahashi, K.; Ogura, M.; Nagasawa, S.; Nakamura, T.; Hirota, H.; Ohta, T.; Nozoe, S. J. Am. Chem. Soc. 1991, 113, 7812-7813.

  Itagaki, F.; Shigemori, H.; Ishibashi, M.; Nakamura, T.; Sasaki, T.; Kobayashi, J. J. Org. Chem. 1992, 57,
- Takao, T.; Yoshino, K.; Suzuki, N.; Shimonishi, Y. Biomed. Environ. Mass Spectrom. 1990, 19, 705-712.
- Mori, K.; Sasaki, M.; Tamada, S.; Suguro, T.; Masuda, S. Tetrahedron 1979, 35, 1601-1605.
- Roepstorff, P; Fohlman, J. Biomed. Mass Spectrom. 1984, 11, 601.
- Gulavita, N. K.; Gunasekera, S. P.; Pomponi, S. A.; Robinson, E. V. J. Org. Chem. 1992, 57, 1767-1772.
- Foster, M. P.; Concepción, G. P.; Garaan, G. B.; Ireland, C. M. J. Org. Chem. 1992, 57, 6671-6675.
- 11. Favert, M. E.; Paschal, J. W.; Elzey, T. K.; Boeck, L. D. J. Antibiot. 1992, 45, 1499-1511.
- 12. Nagai, M.; Ogawa, K.; Muraoka, Y.; Naganawa, H.; Aoyagi, T.; Takeuchi, T. J. Antibiot. 1991, 44, 956-961.
- 13. Ikai, K.; Takesako, K.; Shiomi, K.; Moriguchi, M.; Umeda, Y.; Yamamoto, J.; Kato, I.; Naganawa, H. J. Antibiot. 1991, 44, 925-933.